



Name: _____

Collisions Username: _____

Class: _____

Ionic Bonding Quest

Complete this quest using the Challenge Levels 3, 5 - 10 and the Connected Levels in the Ionic Bonding Game.

MISSION 1. GATHER YOUR INTEL

Use your Collisions gameplay experience to gather the following intel from specific Ionic Bonding levels:

1. Determine the cation in the target.
 - a. Label the element symbol(s) in the large square.
 - b. Write the charge in the smaller, dashed box.
2. Determine the anion in the target.
 - a. Label the element symbol(s) in the large square.
 - b. Write the charge in the smaller, dashed box.

Sample Target

Cation (+):

K

+

Anion (-):

S

2 ⁻

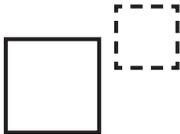
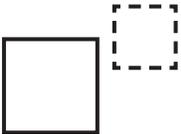
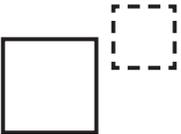
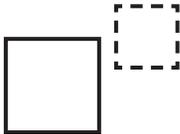
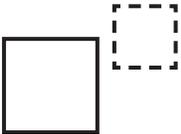
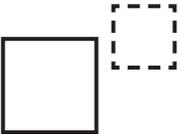
MISSION 2. EXPOSE THE DETAILS

Use your expertise to expose the following information for each target.

Sample Target	
Chemical Formula	K_2S
Compound Name	Potassium Sulfide
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses 1 electron. Each anion gains 2 electrons.
Formula Mass (g/mol)	110.27 g/mol

Ionic Bonding - Challenge Level 3

MISSION 1. GATHER YOUR INTEL

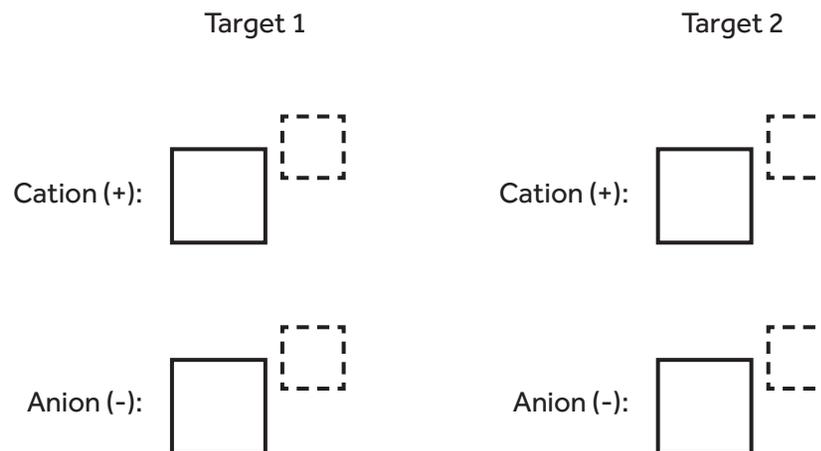
	Target 1	Target 2	Target 3
Cation (+):			
Anion (-):			

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding - Challenge Level 5

MISSION 1. GATHER YOUR INTEL



MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2
Chemical Formula		
Compound Name		
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)		

Ionic Bonding - Challenge Level 6

MISSION 1. GATHER YOUR INTEL

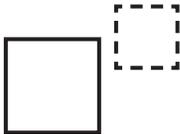
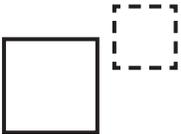
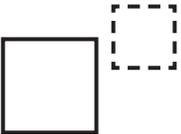
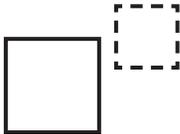
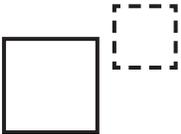
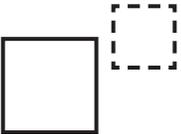
	Target 2	Target 3
Cation (+):	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black; margin-left: 10px;" type="text"/>	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black; margin-left: 10px;" type="text"/>
Anion (-):	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black; margin-left: 10px;" type="text"/>	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black; margin-left: 10px;" type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2
Chemical Formula		
Compound Name		
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)		

Ionic Bonding - Challenge Level 7

MISSION 1. GATHER YOUR INTEL

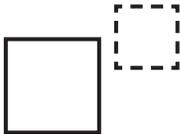
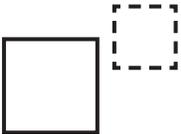
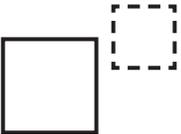
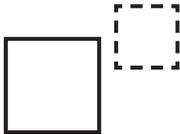
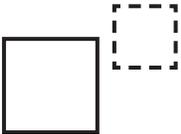
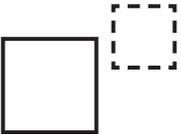
	Target 1	Target 2	Target 3
Cation (+):			
Anion (-):			

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding - Challenge Level 8

MISSION 1. GATHER YOUR INTEL

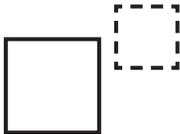
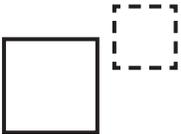
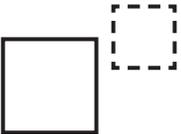
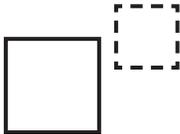
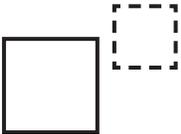
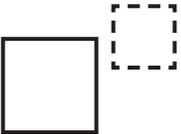
	Target 1	Target 2	Target 3
Cation (+):			
Anion (-):			

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding - Challenge Level 9

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):			
Anion (-):			

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding - Challenge Level 10

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):	<div style="display: flex; align-items: center; justify-content: center;"> <div style="border: 1px solid black; width: 40px; height: 40px; margin-right: 10px;"></div> <div style="border: 1px dashed black; width: 30px; height: 30px; margin-right: 10px;"></div> </div>	<div style="display: flex; align-items: center; justify-content: center;"> <div style="border: 1px solid black; width: 40px; height: 40px; margin-right: 10px;"></div> <div style="border: 1px dashed black; width: 30px; height: 30px; margin-right: 10px;"></div> </div>	<div style="display: flex; align-items: center; justify-content: center;"> <div style="border: 1px solid black; width: 40px; height: 40px; margin-right: 10px;"></div> <div style="border: 1px dashed black; width: 30px; height: 30px; margin-right: 10px;"></div> </div>
Anion (-):	<div style="display: flex; align-items: center; justify-content: center;"> <div style="border: 1px solid black; width: 40px; height: 40px; margin-right: 10px;"></div> <div style="border: 1px dashed black; width: 30px; height: 30px; margin-right: 10px;"></div> </div>	<div style="display: flex; align-items: center; justify-content: center;"> <div style="border: 1px solid black; width: 40px; height: 40px; margin-right: 10px;"></div> <div style="border: 1px dashed black; width: 30px; height: 30px; margin-right: 10px;"></div> </div>	<div style="display: flex; align-items: center; justify-content: center;"> <div style="border: 1px solid black; width: 40px; height: 40px; margin-right: 10px;"></div> <div style="border: 1px dashed black; width: 30px; height: 30px; margin-right: 10px;"></div> </div>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding and Ions - Connected Level 1

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2
Cation (+):	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	Cation (+): <input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>
Anion (-):	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	Anion (-): <input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2
Chemical Formula		
Compound Name		
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)		

Ionic Bonding and Ions - Connected Level 2

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):	<input type="text"/> <input type="text"/>	<input type="text"/> <input type="text"/>	<input type="text"/> <input type="text"/>
Anion (-):	<input type="text"/> <input type="text"/>	<input type="text"/> <input type="text"/>	<input type="text"/> <input type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding and Ions - Connected Level 3

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>
Anion (-):	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px; border: 1px solid black;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding and Acids & Bases - Connected Level 1

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):	<input type="text"/> <input style="border: 1px dashed black;" type="text"/>	<input type="text"/> <input style="border: 1px dashed black;" type="text"/>	<input type="text"/> <input style="border: 1px dashed black;" type="text"/>
Anion (-):	<input type="text"/> <input style="border: 1px dashed black;" type="text"/>	<input type="text"/> <input style="border: 1px dashed black;" type="text"/>	<input type="text"/> <input style="border: 1px dashed black;" type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding and Acids & Bases - Connected Level 2

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>
Anion (-):	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			

Ionic Bonding and Acids & Bases - Connected Level 3

MISSION 1. GATHER YOUR INTEL

	Target 1	Target 2	Target 3
Cation (+):	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>
Anion (-):	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>	<input style="width: 40px; height: 40px;" type="text"/> <input style="width: 40px; height: 40px; border: 1px dashed black;" type="text"/>

MISSION 2. EXPOSE THE DETAILS

	Target 1	Target 2	Target 3
Chemical Formula			
Compound Name			
Indicate the number of electrons lost and gained by each of the atoms in the compound.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.	Each cation loses ____ electrons. Each anion gains ____ electrons.
Formula Mass (g/mol)			